



Sl. No.

Total No. of Pages : 4

V Semester III B.Sc. Examination, March/April - 2023
(Semester Scheme) (CBCS) (2018-19 Batch and Onwards)
(DSE-2A) CHEMISTRY (Paper - V)

Time : 3 Hours

Max. Marks : 80

*Instruction : Write equation and diagram wherever necessary.***Part - A****Answer all the questions :****[8 × 1 = 8]**

1. a) What is anodizing?
- b) Mention one application of carbon fibres.
- c) Give an example for Sulpha drugs.
- d) State isoprene rule.
- e) Give one importance of Vitamin K.
- f) State Grothaus - Draper's law.
- g) What is Chemiluminescence?
- h) Write the selection rule for vibrational spectra.

Part - B**(Inorganic Chemistry)****Answer any three questions :****[3 × 8 = 24]**

2. a) Explain the manufacture and processing of glass. [4]
- b) How is potassium sulphate manufactured? [2]
- c) What are water paints and oil paints? [2]

3. a) Write a note on setting process of cement. [3]
b) Give the composition and properties of Sodalime glass. [3]
c) Mention the different types of Fertilizers. [2]
4. a) Give the steps involved in the manufacture of Ceramics. [2]
b) Write a note on : [4]
i) Fire retardant paints.
ii) Plastic paints.
c) Give any two objectives of surface coatings. [2]
5. a) Discuss the manufacture of calcium ammonium nitrate. [3]
b) Write a note on : [3]
i) Carbon nanotubes.
ii) Enamels.
c) Discuss the manufacture of urea. [2]

Part - C
(Organic Chemistry)

Answer any three questions :

[3 × 8 = 24]

6. a) Give the synthesis of adrenaline. [2]
b) Give any two physical importance of alkaloids. [2]
c) Explain the Skraup's synthesis of Quinoline with mechanism. [4]

7. a) What are terpenes? Write the synthesis of citral. [4]
b) Elucidate the structure of Vitamin A. [4]
8. a) How is paracetamol synthesised? Give an example of Analgesic. [3]
b) Write the structure of testosterone. [2]
c) How is uric acid converted into Caffeine? [3]
9. a) Give the synthesis of thiophene. [2]
b) Write the structural formula of calciferol. [2]
c) How do you show the presence of i) Two 3° nitrogen atoms ii) Pyridine ring in nicotine. <https://www.uomonline.com> [4]

Part - D
(Physical Chemistry)

Answer any three questions :

[3 × 8 = 24]

10. a) Derive the mathematical form of Lambert -Beer's law. [3]
b) Write the vibrational modes of CO₂ molecule. [3]
c) Mention the applications of Raman spectra. [2]
11. a) Derive the expression for moment of inertia of a diatomic molecule. [4]
b) Write a note on photosensitization. [2]
c) Mention the applications of IR spectroscopy. [2]

12. a) Give reasons for low and high quantum yield. [4]
- b) Explain radiolysis of acetic acid. [2]
- c) Write a note on Franck - Condon principle. [2]
13. a) Write a note on uranyl oxalate actinometer. [3]
- b) How are Stoke's and antistoke's lines formed in Raman spectrum? Explain. [3]
- c) Give any two comparison between radiation chemistry and photochemistry. [2]

